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39 E On this test, the easiest way to balance an equation is by plugging in the answers. The test writer tells you that the coefficient in front of H_2O is 6, so put that in there. Now, see which of the answer choices, if placed in front of NF_3 , would result in a balanced equation. If, for instance, we try choice (B), we'd have 1 mole of N on the left, which would give us 3 moles of F on the left.

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Answers 1. $1 \text{ SnO}_2 + 2 \text{ H}_2 ?$
 $1 \text{ Sn} + 2 \text{ H}_2 \text{ O}$ 2. $3 \text{ KOH} + 1$

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H₃PO₄ ? 1 K₃PO₄ + 3 H₂O
3. 2 KNO₃ + 1 H₂CO₃
? 1 K₂CO₃ + 2 HNO₃
4. 1 Na₃PO₄ + 3 HCl ? 3 NaCl +
1 H₃PO₄
5. 1 TiCl₄ + 2 H₂O ? 1 TiO₂ + 4 HCl
6. 1 C₂H₆O + 3 O₂ ? 2 CO₂ + 3
H₂O
7. 2 Fe + 6 HC₂H₃O₂ ? 2 Fe(C₂H₃O₂)₃ + 3
H₂
8. 4 NH₃ + 5 O₂ ? 4 NO + 6 H₂O
9. 1 B₂Br₆ + 6 HNO₃ ? 2 B(NO₃)₃ + 6 HBr
10. 4 NH₄OH + 1 K₂(SO₄)₂ · 12H₂O ? 1 Al(OH)₃ + 2
(NH₄)₂SO₄ ...

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Chemistry is a very fun subject and being a 10th grader it is important for you to ensure that you understand each topic taught before the final quiz. The quiz below is designed to see just how attentive you were in class this past semester, give it a shot and see just how much you remember so far. All the best!

Quiz: Chemistry For Grade 10 Student - ProProfs Quiz

This document gives detailed answer explanations to chemistry practice questions

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from The SAT Subject Tests™ Student Guide. By reviewing them, you'll get to know the types of questions on the test and learn your strengths and weaknesses. Estimated difficulty level is based on a 1-5 scale, with 1 the easiest and 5 the most difficult. For more about the SAT Subject Tests, go to satsubjecttests.org.
1.

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